

## Problem 6: The hydrogen atom (10 points)

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The figure below shows the radial function  $R_{n,\ell}(r)$  for a stationary state of atomic hydrogen. The normalized Hamiltonian eigenfunction for this state, in atomic units, is

$$\psi_{n,\ell,m_\ell}(\mathbf{r}) = \frac{1}{81} \sqrt{\frac{2}{\pi}} (6-r) e^{-r/3} \cos \theta. \quad (1)$$

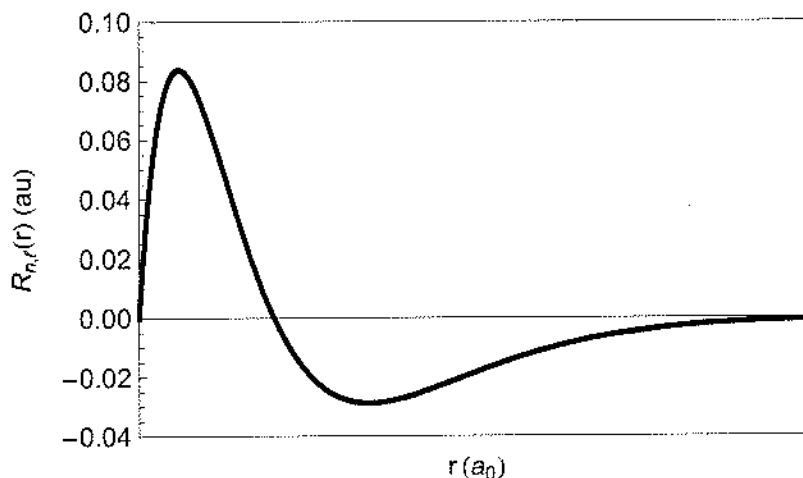


Figure 1: A radial function for a stationary state of atomic hydrogen.

- ~~(a)~~ 1. **3 points.** What are the values of the quantum numbers  $n$ ,  $\ell$ , and  $m_\ell$  for this state? To receive any credit, you must fully justify your answer.
- ~~(b)~~ 2. **1 points.** What is the energy (in eV) of this state?
- ~~(c)~~ 3. **2 points.** What are the mean value and uncertainty in  $r$  (in atomic units) for this state?
- ~~(d)~~ 4. **2 points.** Calculate the value of  $r$  (in atomic units) at which a position measurement would be most likely to find the electron if the atom is in this state.
- ~~(e)~~ 5. **2 points.** From Eq. 1, generate the normalized eigenfunction  $\psi_{n,\ell,m_\ell+1}(\mathbf{r})$ .

**Hint:**

$$\int_0^\infty e^{-2r/3} r^n dr = n! \left(\frac{3}{2}\right)^{n+1} \quad (2)$$

**Hint:** The following table gives the orbital-angular-momentum operators in Cartesian and spherical coordinates.

Component	Cartesian coordinates	Spherical coordinates
$\hat{L}_x$	$-i\hbar \left( y \frac{\partial}{\partial z} - z \frac{\partial}{\partial y} \right)$	$i\hbar \left( \sin \varphi \frac{\partial}{\partial \theta} + \cot \theta \cos \varphi \frac{\partial}{\partial \varphi} \right)$
$\hat{L}_y$	$-i\hbar \left( z \frac{\partial}{\partial x} - x \frac{\partial}{\partial z} \right)$	$-i\hbar \left( \cos \varphi \frac{\partial}{\partial \theta} - \cot \theta \sin \varphi \frac{\partial}{\partial \varphi} \right)$
$\hat{L}_z$	$-i\hbar \left( x \frac{\partial}{\partial y} - y \frac{\partial}{\partial x} \right)$	$-i\hbar \frac{\partial}{\partial \varphi}$
$\hat{L}^2$	$\hat{L}_x^2 + \hat{L}_y^2 + \hat{L}_z^2$	$-\hbar^2 \left[ \frac{1}{\sin \theta} \frac{\partial}{\partial \theta} \left( \sin \theta \frac{\partial}{\partial \theta} \right) + \frac{1}{\sin^2 \theta} \frac{\partial^2}{\partial \varphi^2} \right]$

Table 1: Components and square of the orbital angular momentum operator in Cartesian and spherical coordinates.

(a)

In general, we know  $l = 0, \dots, n-1$  and  $m_l = -l, \dots, l$ .

Since there is no  $\phi$  dependence, we know  $\boxed{m_l = 0}$ . Our Legendre polynomial is  $P = \cos\theta$ , which corresponds to  $m_l = 0$  and  $\boxed{l = 1}$ . So we know we must have  $n > 1$ .

In general, the exponential portion of the radial equation is given by

$$e^{-x/2},$$

where

$$x = \frac{Zr}{a_0}.$$

So

$$\begin{aligned} e^{-x/2} &= e^{-\frac{1}{2} \left( \frac{Zr}{na_0} \right)} \\ &= e^{-r/na_0}. \end{aligned}$$

In atomic units,  $a_0 = 1$ , so since our exponential portion is

$$e^{-r/3},$$

we must have  $\boxed{n=3}$ . Thus,

$$|n \ l \ m_l\rangle = |3 \ 1 \ 0\rangle.$$

(b)

The energy is given by

$$E_n = \frac{E_1}{n^2},$$

where

$$E_1 = -13.6 \text{ eV}.$$

So since  $n=3$ , we have

$$\begin{aligned} E_3 &= \frac{-13.6 \text{ eV}}{(3)^2} \\ &= \frac{-13.6 \text{ eV}}{9} \end{aligned}$$

$$E_3 \approx -1.51 \text{ eV}$$

(c)

We want to determine  $\langle r \rangle$  and  $\Delta r$ . So

$$\begin{aligned}
 \langle r \rangle &= \int_0^{2\pi} \int_0^\pi \int_0^\infty \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} \right)^2 r (b-r)^2 e^{-2r/3} \cos^2 \theta r^2 \sin \theta dr d\theta d\phi \\
 &= \frac{4\pi}{3} \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} \right)^2 \int_0^\infty (r^5 - 12r^4 + 36r^3) e^{-2r/3} dr \\
 &= \frac{8}{3^9} \left[ \frac{5!}{(2/3)^6} - 12 \frac{4!}{(2/3)^5} + 36 \frac{3!}{(2/3)^4} \right] \\
 &= \frac{8}{3^9} \left[ \frac{5 \cdot 4 \cdot 3 \cdot 2 \cdot 3^6}{2^2 \cdot 2 \cdot 2^3} - 12 \frac{4 \cdot 3 \cdot 2 \cdot 3^5}{2^2 \cdot 2 \cdot 2^2} + 36 \frac{3 \cdot 2 \cdot 3^4}{1 \cdot 2^3} \right] \\
 &= \frac{2^3}{3^9} \left[ \frac{5 \cdot 3^7}{2^3} - \frac{2^4 \cdot 3^7}{2^4} + \frac{2^2 \cdot 3^7}{2^3} \right] \\
 &= \frac{1}{3^9} [5 \cdot 3^7 - 2^3 \cdot 3^7 + 2^2 \cdot 3^7] \\
 &= \frac{5}{3^2} - \frac{2^3}{3^2} + \frac{2^2}{3^2} \\
 &= \frac{5}{9} - \frac{8}{9} + \frac{4}{9}
 \end{aligned}$$

$$\boxed{\langle r \rangle = \frac{1}{9}}$$

(c), cont'd...

We know

$$\Delta r = \sqrt{\langle r^2 \rangle - \langle r \rangle^2},$$

so we need to determine  $\langle r^2 \rangle$ . Then

$$\begin{aligned}\langle r^2 \rangle &= \frac{2}{3} \cdot 2\pi \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} \right)^2 \int_0^\infty \left[ (r^6 - 12r^5 + 36r^4) e^{-2r/3} dr \right] \\&= \frac{4\pi}{3} \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} \right)^2 \left[ \frac{6!}{(2/3)^7} - 12 \frac{5!}{(2/3)^6} + 36 \frac{4!}{(2/3)^5} \right] \\&= \frac{8}{3^9} \left[ \frac{6 \cdot 5 \cdot 4 \cdot 3 \cdot 2 \cdot 3^7}{2 \cdot 2^2 \cdot 2 \cdot 2^3} - \frac{5 \cdot 4 \cdot 3 \cdot 2 \cdot 2^2 \cdot 3 \cdot 3^6}{2^2 \cdot 2 \cdot 2^4 \cdot 2} + \frac{4 \cdot 3 \cdot 2 \cdot 3^2 \cdot 2^2 \cdot 3^5}{2^2 \cdot 2 \cdot 2^2} \right] \\&= \frac{2^3}{3^9} \left[ \frac{3 \cdot 5 \cdot 3 \cdot 3^7}{2^3} - \frac{5 \cdot 3 \cdot 3 \cdot 3^6}{2} + 3 \cdot 3^2 \cdot 3^5 \right] \\&= \frac{2^3}{3^9} \left[ \frac{5 \cdot 3^9}{2^3} - \frac{5 \cdot 3^8}{2} + 3^8 \right] \\&= 5 - \frac{2 \cdot 5}{3} + \frac{2^3}{3} \\&= 5 - \frac{20}{3} + \frac{8}{3} \\&= 1\end{aligned}$$

and thus,

$$\Delta r = \sqrt{1 - (1/9)^2}$$

$$= \sqrt{\frac{80}{81}}$$

$$\boxed{\Delta r = \frac{4\sqrt{5}}{9}}$$

(d)

We want to determine the most probable radius. The radial probability density is

$$\begin{aligned} \rho &= 4\pi r^2 |\psi_r|^2 \\ &= 4\pi r^2 \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} \right)^2 (r^2 - 12r + 36) e^{-2r/3} \end{aligned}$$

We want to maximize this. So

$$\begin{aligned} \frac{d\rho}{dr} &= \frac{d}{dr} \left[ \frac{8}{81^2} (r^4 - 12r^3 + 36r^2) e^{-2r/3} \right] = 0 \\ &= \frac{8}{81^2} \left[ -\frac{2}{3} e^{-2r/3} (r^4 - 12r^3 + 36r^2) \right. \\ &\quad \left. + e^{-2r/3} (4r^3 - 36r^2 + 72r) \right] = 0 \end{aligned}$$

$$+\frac{2}{3} (r^4 - 12r^3 + 36r^2) = 4r^3 - 36r^2 + 72r$$

$$r^4 - 12r^3 + 36r^2 = 6r^3 - 54r^2 + 108r$$

$$r^3 - 12r^2 + 36r = 6r^2 - 54r + 108$$

$$r^3 - 18r^2 + 90r - 108 = 0$$

The roots are

$$r = -3(\sqrt{2} - 2), 3(\sqrt{2} + 2), 6$$

Based on the plot, it seems as though our most probable radius is

$$r = 3(2 - \sqrt{2})$$

$$\boxed{r = 6 - 3\sqrt{2}}$$

which is verified by plugging each of our choices into  $\psi_r$ .

(e)

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In order to find  $\Psi_{n,l,m_l+1}$ , we want to apply the raising operator

$$L_+ = L_x + iL_y.$$

We have

$$L_+ = i\hbar \left( \sin\phi \frac{\partial}{\partial\theta} + \cot\theta \cos\phi \frac{\partial}{\partial\phi} \right) + \hbar \left( \cos\phi \frac{\partial}{\partial\theta} - \cot\theta \sin\phi \frac{\partial}{\partial\phi} \right)$$

Since  $\Psi_{n,l,m_l}$  does not depend on  $\phi$ , we can eliminate those derivatives.  
Then

$$L_+ = i\hbar \sin\phi \frac{\partial}{\partial\theta} + \hbar \cos\phi \frac{\partial}{\partial\theta}.$$

We want to determine

$$\Psi_{n,l,m_l+1} = L_+ \Psi_{n,l,m_l}.$$

So

$$\begin{aligned} L_+ \Psi_{n,l,m_l} &= \left( i\hbar \sin\phi \frac{\partial}{\partial\theta} + \hbar \cos\phi \frac{\partial}{\partial\theta} \right) \left( \frac{1}{81} \sqrt{\frac{2}{\pi}} (b-r) e^{-r/3} \cos\theta \right) \\ &= \frac{\hbar}{81} \sqrt{\frac{2}{\pi}} \left( i \sin\phi \frac{\partial}{\partial\theta} + \cos\phi \frac{\partial}{\partial\theta} \right) \left( (b-r) e^{-r/3} \cos\theta \right) \\ &= \frac{\hbar}{81} \sqrt{\frac{2}{\pi}} \left[ -i \sin\theta \sin\phi (b-r) e^{-r/3} - \sin\theta \cos\phi (b-r) e^{-r/3} \right] \\ &= \frac{-\hbar}{81} \sqrt{\frac{2}{\pi}} (b-r) e^{-r/3} \left[ \sin\theta (\cos\phi + i \sin\phi) \right] \end{aligned}$$

$$\boxed{\Psi_{n,l,m_l+1} \left( \frac{r}{b} \right) = -\frac{\hbar}{81} \sqrt{\frac{2}{\pi}} (b-r) e^{-r/3} \sin\theta e^{i\phi}}.$$