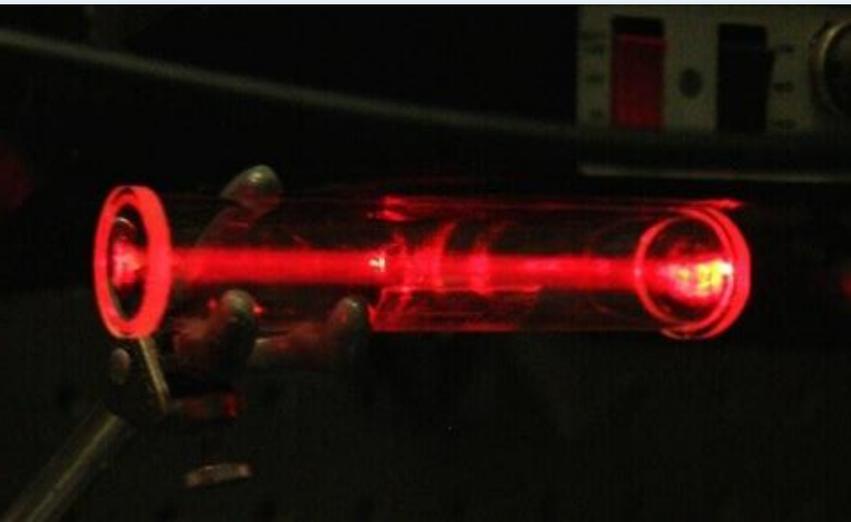


Laser Spectroscopy and Potential Modeling

Kit Leonard

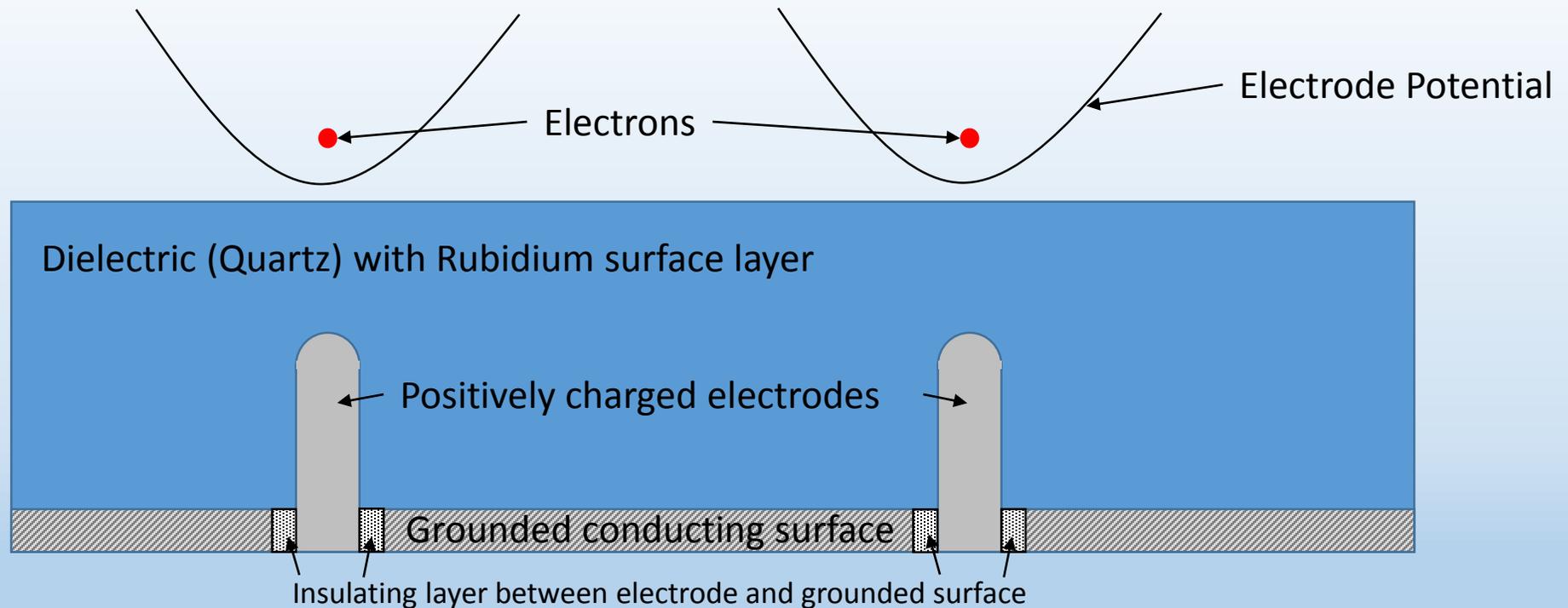
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University of Oklahoma



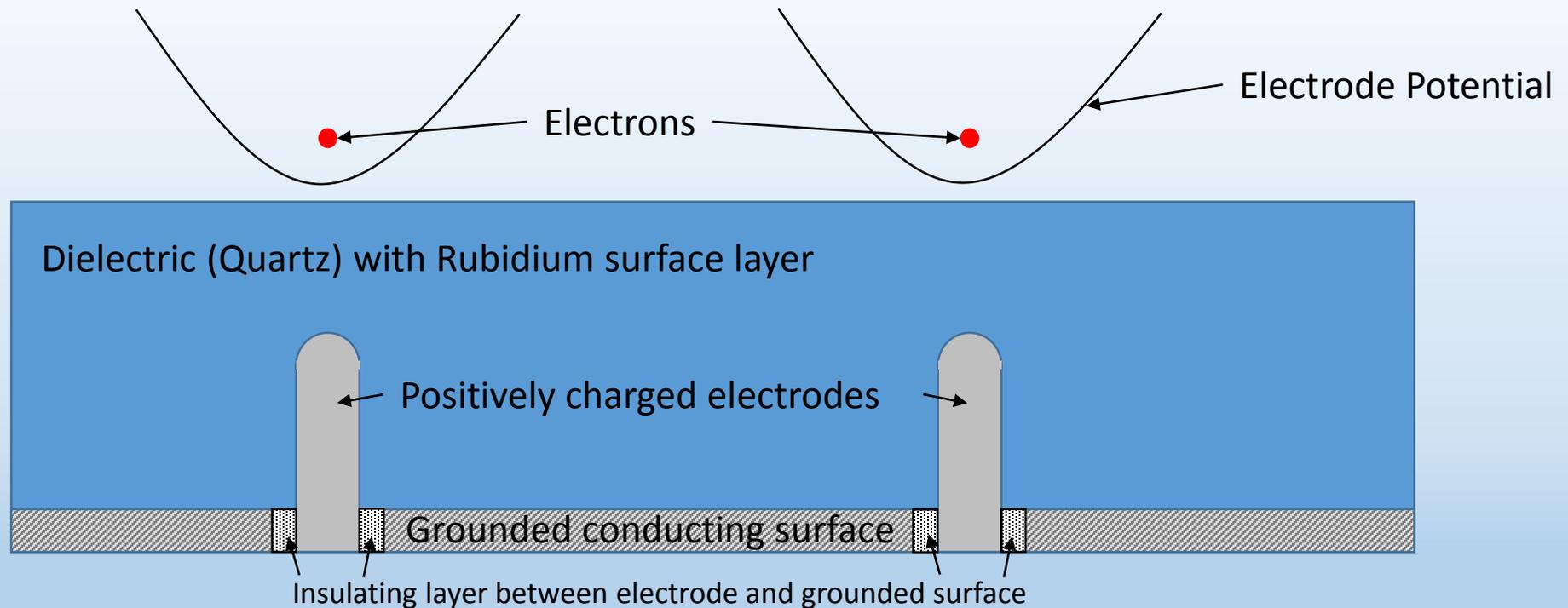
Main Experiment

- Electrons held in free space above a dielectric surface
 - Motion normal to surface quantized by Coulomb potential
 - In-plane motion controlled by electrodes, quantized by harmonic oscillator potential
 - Individual electrons can be excited using an infrared laser
- Goal: Create entangled states that can be used to store quantum information



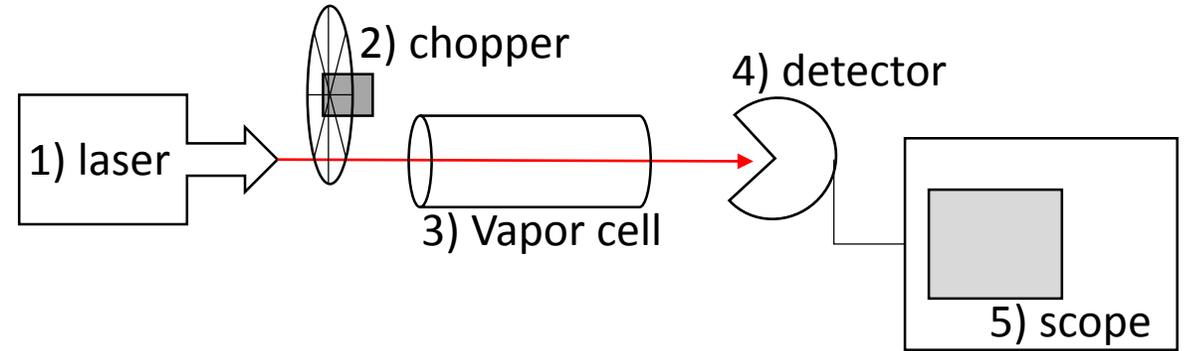
My Projects

- Laser Spectroscopy
 - Tune the laser to excitation frequencies of electrons
 - Precise enough to excite energies normal to surface, avoid in-plane excitation
- Potential Modeling
 - Find parameters for electrodes that produce desired potential shape
 - Energy level spacing/Number of bound states
 - Check accuracy of harmonic oscillator approximation



Spectroscopy Setup

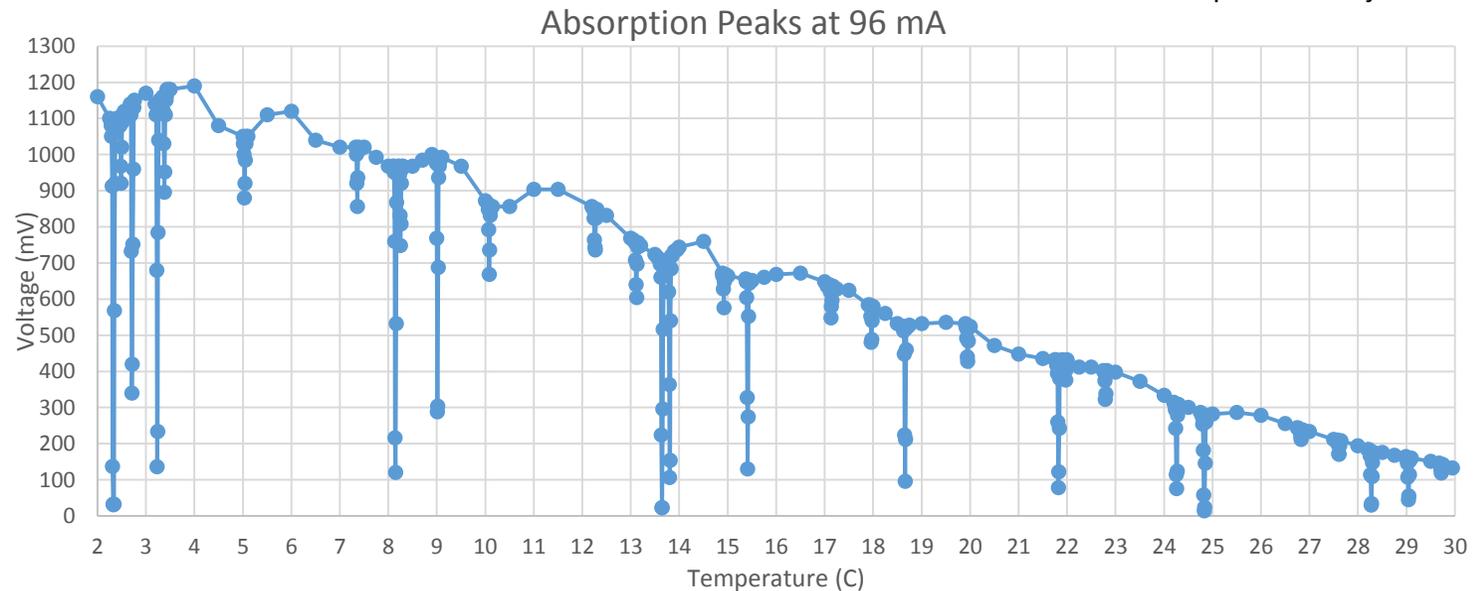
- Measure absorption frequencies of CO₂ using a tunable laser
- Hold current at 96mA, change temperature to adjust frequency
- Drops in output power indicate absorption by CO₂



Experimental Setup

- 1) Quantum Cascade Laser*, Wavelength 4300-4350 nm
- 2) Optical chopper- “chops” laser signal into pulses
- 3) CO₂ vapor cell
- 4) Photodetector- measures incoming laser power, outputs voltage
- 5) Oscilloscope- displays signal from photodetector

*Current and Temperature Adjustable

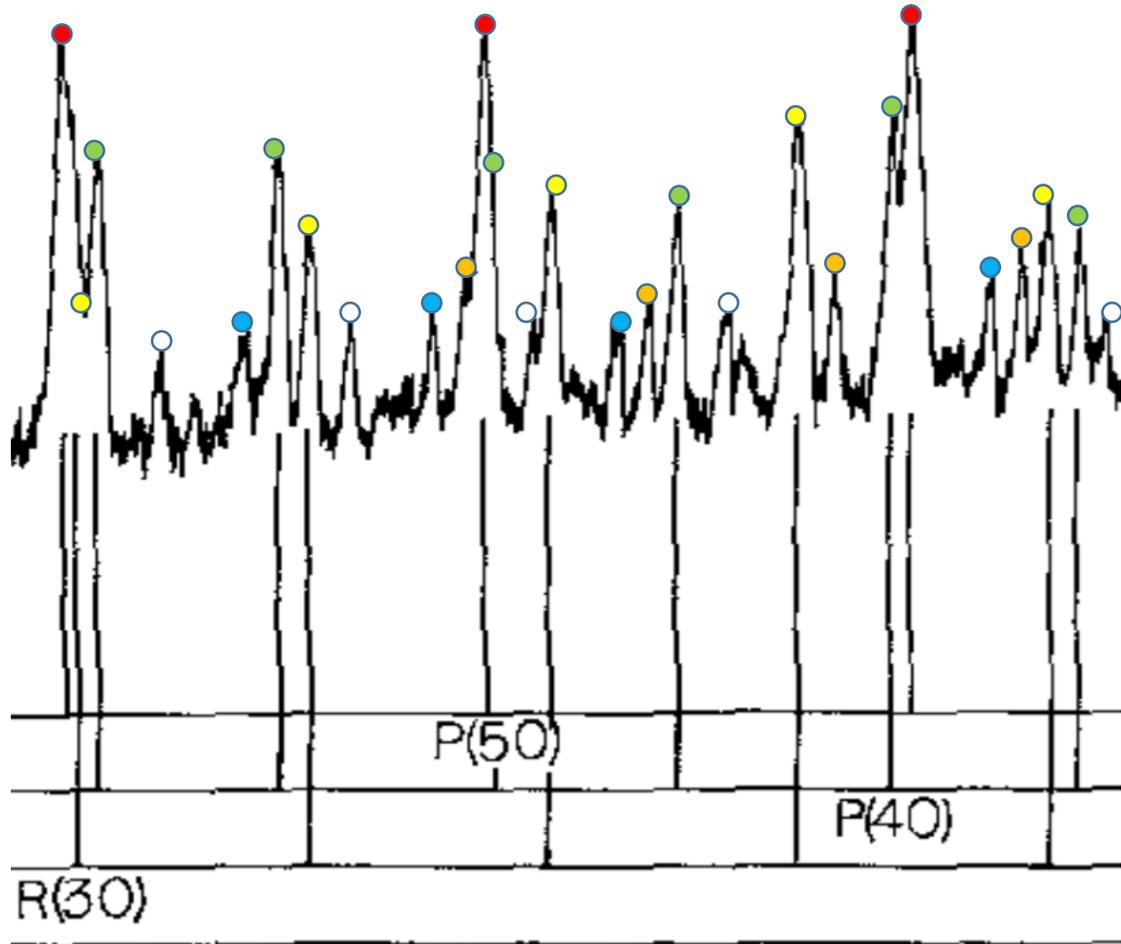


Spectroscopy Objectives

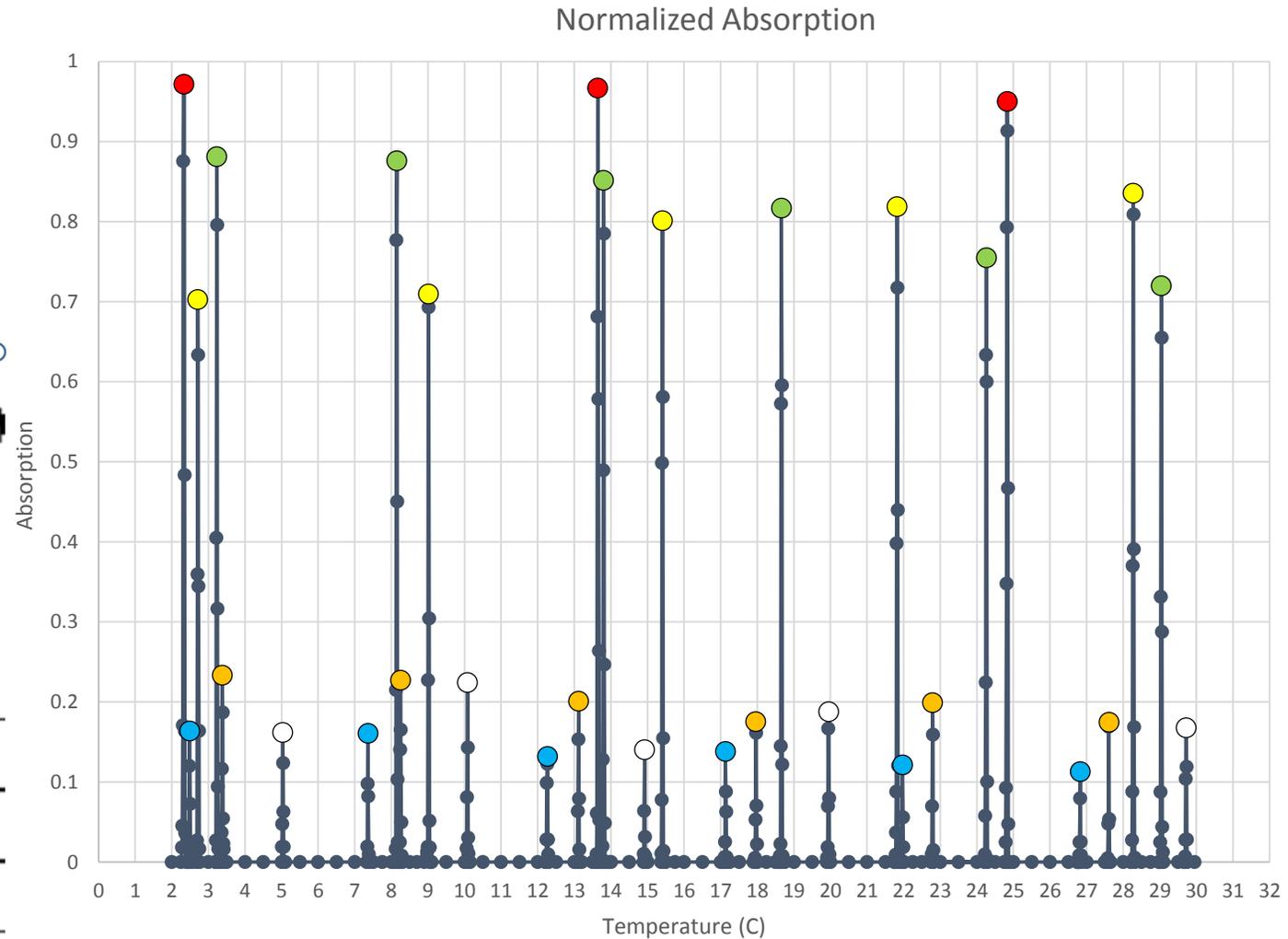
- Match observed absorption pattern to known spectrum of CO₂
 - Frequencies of absorption peaks are well known
- Observe linearity of laser response
 - Linear frequency response needed to match results
 - Linear power response less important, observe general behavior

Results: Absorption Spectrum

Comparison: 1968 Spectroscopy study using spectrographs

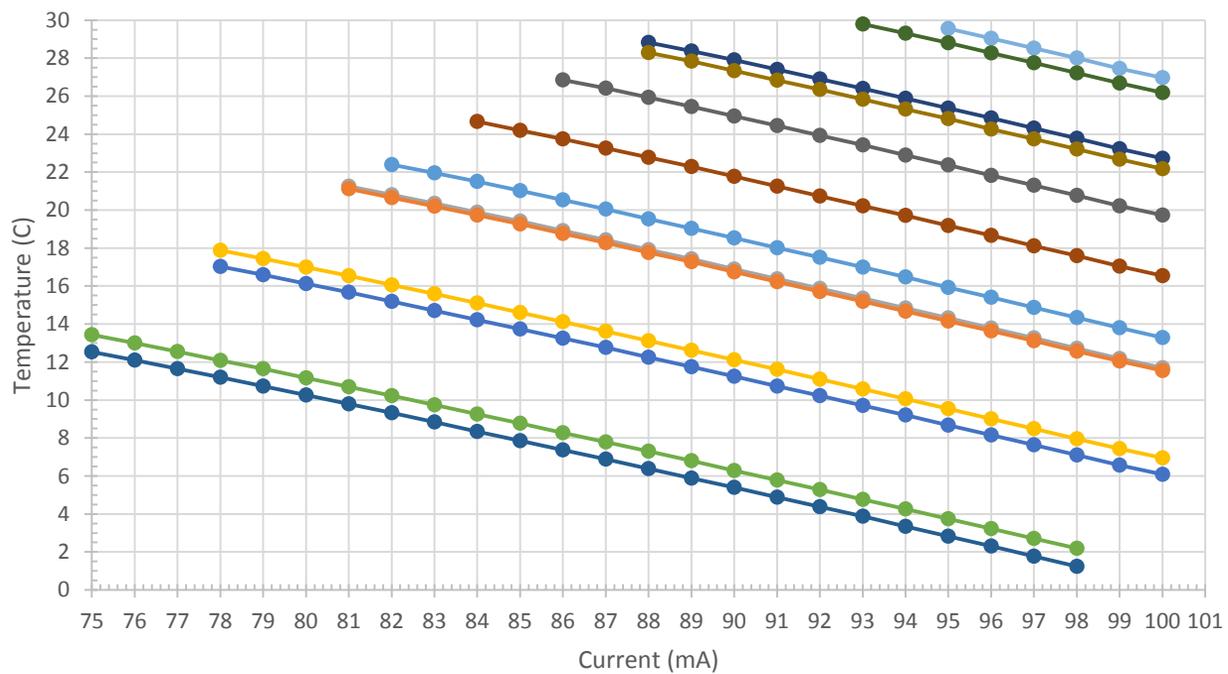


Results obtained through laser spectroscopy

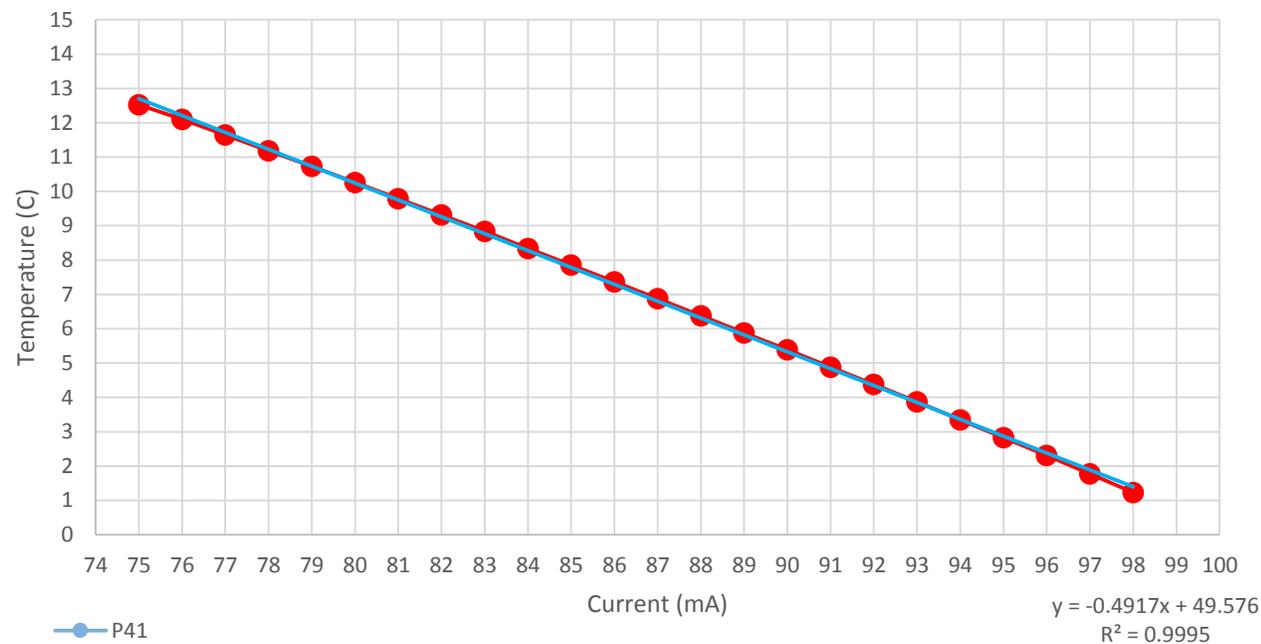


Results: Laser Behavior

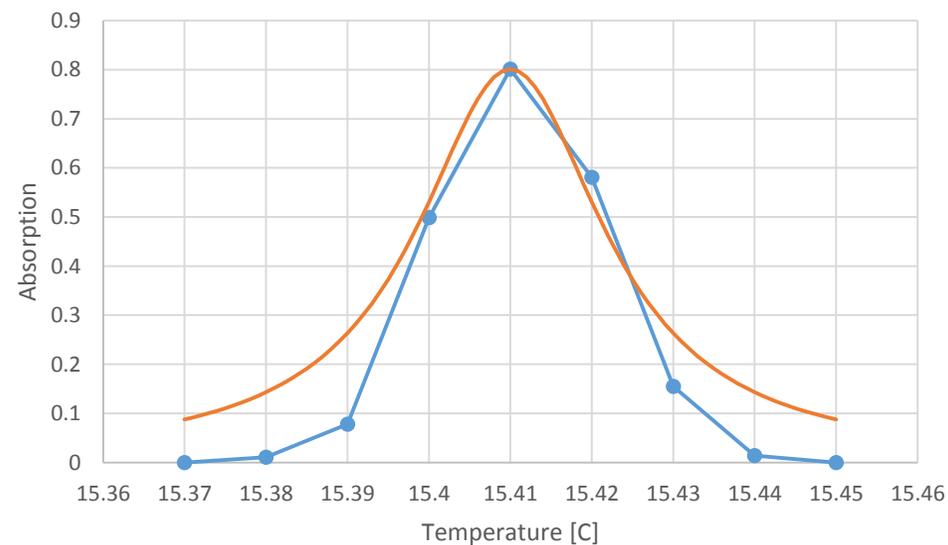
Combined Absorption Conditions



Conditions for P48 Absorption



R26 Absorption vs Lorentzian



- P41
- R22
- P52
- P40
- R24
- P39
- R26
- P38
- P50
- R28
- P37
- P36
- P48

Conclusions: Laser Spectroscopy

- Laser frequency varies linearly and can be accurately determined for given input conditions
- Laser power varies approximately linearly with some fluctuation
- Some broadening of absorption peaks is present, effects are minimal
 - Broadening independent of power
- Laser can be tuned precisely to excite normal-to-surface electron energies while avoiding excitation of in-plane energies

Potential Modeling

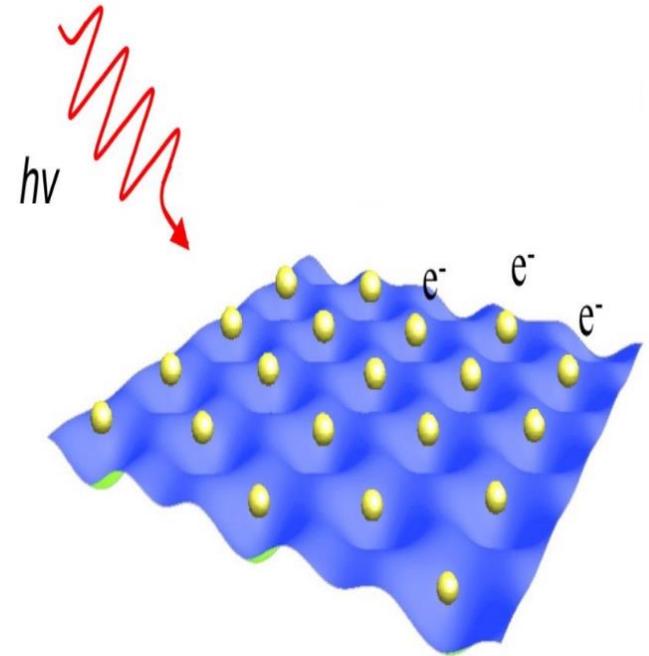
Examine how in-plane motion of electrons is quantized

Find best parameter values for electrodes

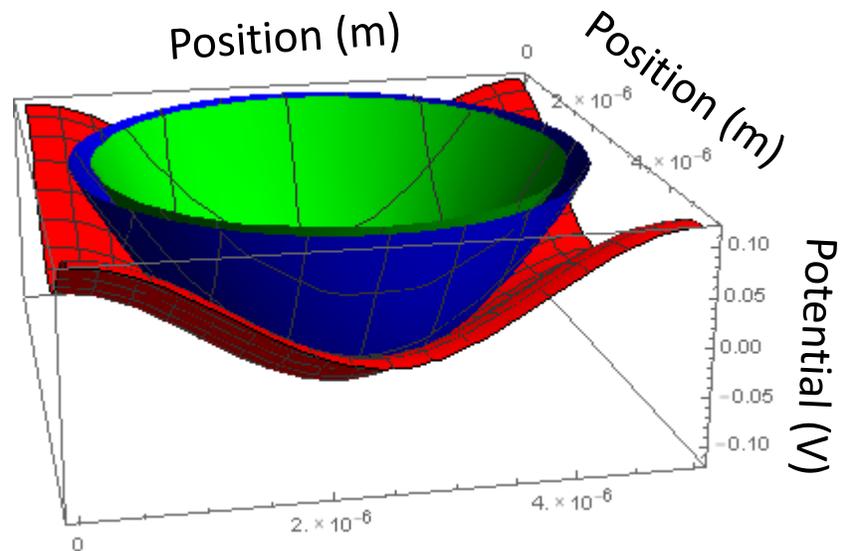
- Similar experiments use a liquid Helium surface to vertically confine electrons
 - Electron spacing $\sim 0.5 \mu\text{m}$, Laser wavelength $\sim 1 \text{ cm}$
- Our experiment uses a dielectric surface
 - Electron spacing $\sim 5 \mu\text{m}$, Laser wavelength $\sim 4.3 \mu\text{m}$

Can electrode geometries from liquid He experiments also be used in our experiment?

How can the electrode potential function be simplified by approximations?



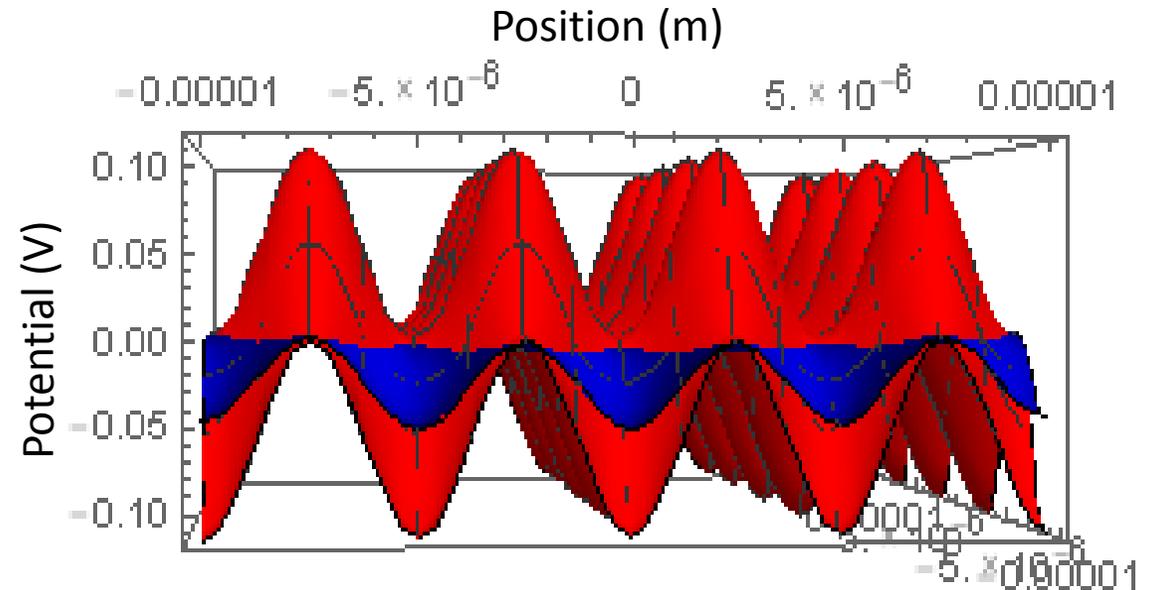
Results



Green: Actual potential (single electrode)

Blue: Harmonic oscillator potential

Red: Sinusoidal multielectrode approximation

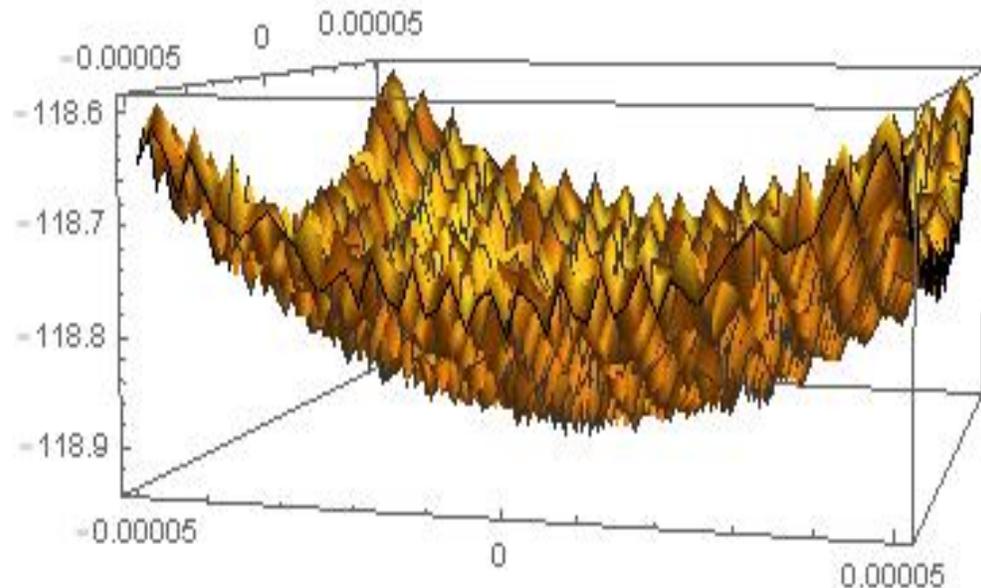


Red: Sinusoidal multielectrode approximation

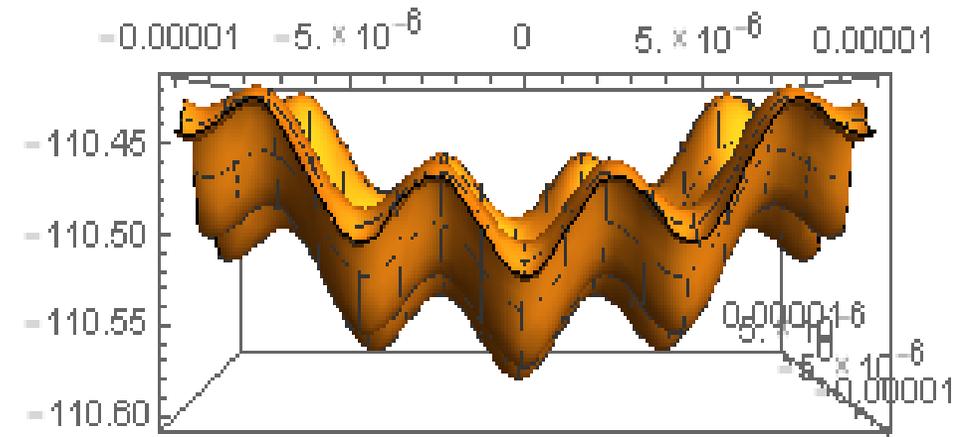
Blue: Actual potential (1000 electrode array)

- +Harmonic Oscillator approximation is valid
- +Electrons can be localized by electrode potentials with many (100+) bound states
- +Bound states are spaced by 20 GHz
- +Sinusoidal approximation* holds when multiplied by a constant
- *comes from a different electrode geometry

Limitations



Zoomed-out view of the actual electrode potential for a 100x100 array. Note that wells have constant depth despite overall curvature.



Potential of a 40x40 array of electrodes. Curvature effects are sharper, well depth is unchanged.

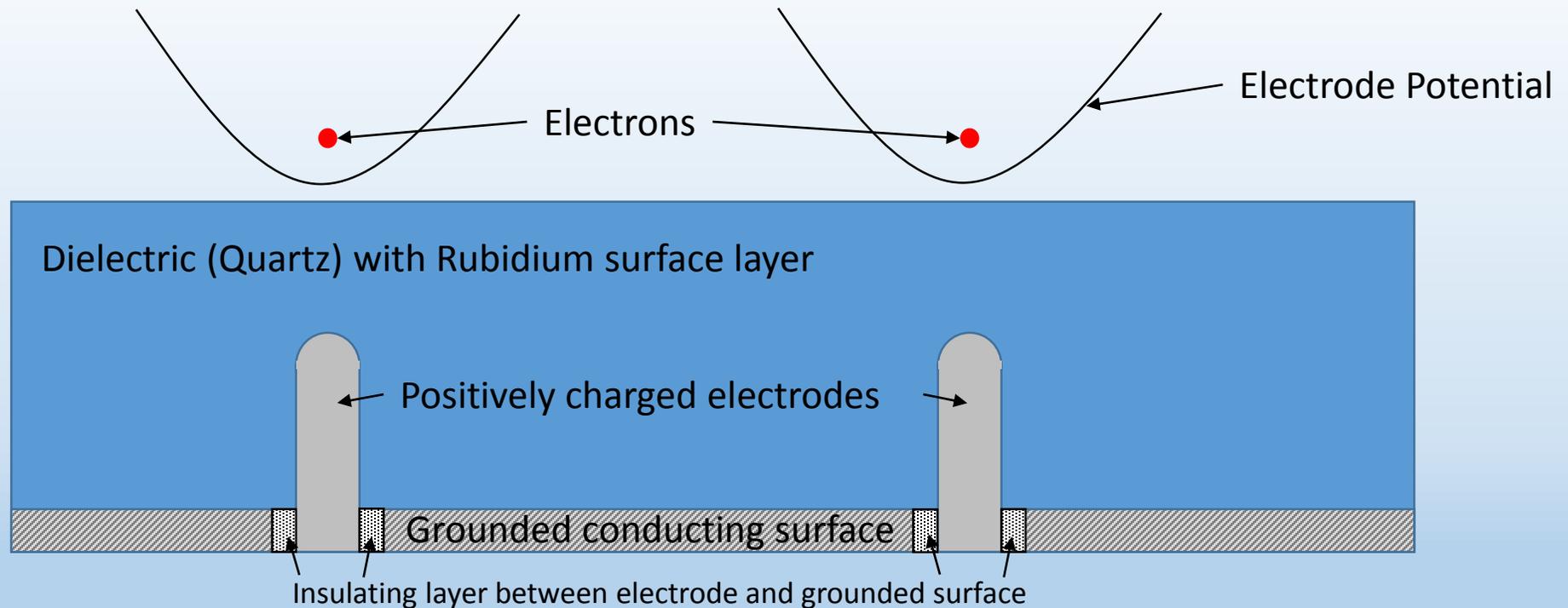
- Sinusoidal approximation not valid over large arrays
 - +actual potential can be flattened out
- Large arrays are needed to see agreement between sinusoidal model and actual potential
- Adjusting parameters has a limited effect on improving fit

Conclusions: Potential Modeling

- In-plane position of electrons can be localized using electrodes
- Electrode potential can be approximated by a harmonic oscillator potential
 - Energy gap between in-plane excited states is ~ 20 GHz as desired
- Large electrode arrays are needed to reduce curvature effects
- Altering electrode dimensions has a limited effect on the shape of the resulting potential

Summary

- Laser is capable of precisely exciting individual electrons
- Electrons can be confined by an electrode array with reasonable dimensions
- Energy gap between in-plane excited states is large enough to avoid unwanted excitations by the laser
- System could potentially be used for quantum information and quantum computing!



Acknowledgements

Dykman, M. I.; Platzman, P.M.; Seddighrad, P.; *Qubits with electrons on liquid helium*, Physical Review B, Vol 67, 2003.

Oberly, Ralph; Rao, K. Narahari; Hahn, Y. H.; McCubbin, T. K.; *Bands of Carbon Dioxide in the Region of 4.3 Microns*, Journal of Molecular Spectroscopy, Vol 25, Issue 2, p. 138-165, 1968.

